

# A Mass Spectrometric Criterion for Determining the B- and E-Ring Hydroxylation Pattern in Dimeric Biflavonoids

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Since only the ions with  $m/z$  corresponding to retro-Diels–Alder fragmentation products consisting of the original A-, B-, C-, and D-rings can be observed by the low-resolution mass spectra of the methyl ether diacetates of biflavonoids these ions can be used to determine the hydroxylation patterns of the B- and E-rings of unknown dimeric biflavonoids provided  $^1\text{H}$  n.m.r. data are available as well.

The chemistry of the dimeric procyanidins is well understood.<sup>1–3</sup> In the past few years significant work has also been devoted to other biflavonoids.<sup>4–6</sup> Methyl ether acetates of biflavonoids have in the past proved to be useful derivatives to allow the differentiation between the alternatives of C(4)–C(6') and C(4)–C(8'') interflavanoid links by  $^1\text{H}$  n.m.r. spectroscopy,<sup>3,7</sup> to allow good mass spectra to be obtained, and to allow for a purification after each derivatisation step.<sup>5</sup> We here show that these derivatives have the further advantage that by interpretation of their mass and  $^1\text{H}$  n.m.r. spectra the hydroxylation pattern of the B- and E-rings can be deduced, a feature up to now not possible with  $^1\text{H}$  n.m.r. spectroscopy, but only possible through degradations of the pure phenolic substances and chromatography of the resultant products.<sup>2</sup>

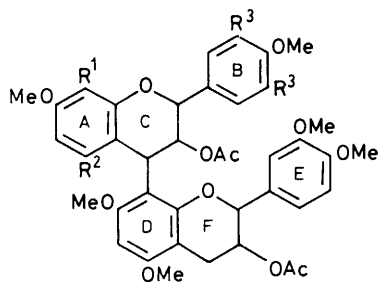
Thus, in the case of derivatives of prodelphinidin B3 (1), a second biflavonoid (2) isolated from beer,<sup>8,9</sup> and also the five

synthetic biflavonoids<sup>5</sup> (3)–(7), only the ions with  $m/z$  values corresponding to retro-Diels–Alder fragmentation products<sup>1</sup> consisting of the original A-, B-, C-, and D-rings can be observed. Indeed, if retro-Diels–Alder fragmentation in the 'lower' flavan unit (after a loss of acetic acid in the 'upper' unit) of (1) and (2) is detectable, an ion with  $m/z$  522 is expected. If the opposite breakdown process (loss of acetic acid in the 'lower' unit, retro-Diels–Alder fragmentation in the 'upper' unit) is detectable, an ion with  $m/z$  492 will be observed. Since only ions with  $m/z$  522 were observed, we concluded that only the retro-Diels–Alder fragmentation products consisting of the original A-, B-, C-, and D-rings are detectable. In the case of the synthetic biflavonoid derivatives<sup>5</sup> (3)–(7) we found ions at  $m/z$  492, 492, 462, 492, and 462, respectively, all indicative of retro-Diels–Alder fragmentation ions as in the case of (1) and (2). If the alternative breakdown pattern were to be detectable we should have observed peaks at  $m/z$  462, 462, 492, 462, and 492, respectively. This fragmentation probably occurs as well since the retro-Diels–Alder fragments consisting of the original B-ring are found in the case of (3),  $m/z$  252; (4), 252; (5), 192; (6), 252; and (7), 192. Since in the spectra of (1), (2), (5), (7) the ions with  $m/z$  222 (corresponding to the original E-rings) are found, these ions can be of no diagnostic value.

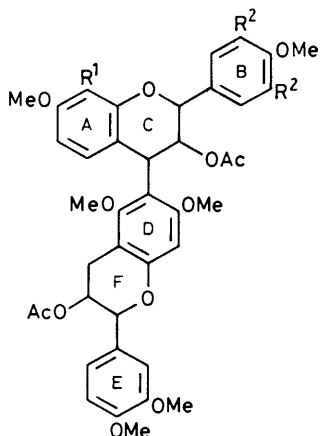
From the above it is clear that, for example, a biflavonoid such as prodelphinidin B3 can be distinguished from a dimer having (+)-catechin as 'upper' unit and (+)-gallocatechin as the 'terminal' unit by mass spectrometry of the methyl ether diacetate derivative. It is further obvious that  $^1\text{H}$  n.m.r. spectroscopy is less adequate in this respect.

Helpful discussions with Professor P. Dondeyne are appreciated.

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- (1)  $R^1 = \text{H}$ ,  $R^2 = R^3 = \text{OMe}$ , 2*R*,3*S*,4*S*; 2*R*,3*S*  
 (2)  $R^1 = \text{H}$ ,  $R^2 = R^3 = \text{OMe}$ , 2*R*,3*R*,4*S*; 2*R*,3*S*  
 (3)  $R^1 = R^2 = \text{H}$ ,  $R^3 = \text{OMe}$ , 2*R*,3*S*,4*R*; 2*R*,3*S*  
 (4)  $R^1 = R^2 = \text{H}$ ,  $R^3 = \text{OMe}$ , 2*R*,3*S*,4*S*; 2*R*,3*S*  
 (5)  $R^2 = R^3 = \text{H}$ ,  $R^1 = \text{OMe}$ , 2*R*,3*R*,4*R*; 2*R*,3*S*



- (6)  $R^1 = \text{H}$ ,  $R^2 = \text{OMe}$ , 2*R*,3*S*,4*S*; 2*R*,3*S*  
 (7)  $R^2 = \text{H}$ ,  $R^1 = \text{OMe}$ , 2*R*,3*R*,4*R*; 2*R*,3*S*

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